

AHMADU BELLO UNIVERSITY

SCIENCE POST UTME PAST QUESTIONS

PrepsNG Learning Center

ABU ZARIA POST UTME PAST QUESTIONS FOR SCIENCE

2017/2018 POST-UTME SCREENING EXERCISE

Time allowed: 1 hour

Read the following instructions carefully:

- 1. Use HB pencil to shade your answers. Ensure that any shading in error is thoroughly erased.
- 2. Candidates should indicate the question Paper Type given to them in the appropriate space in the

Answer Sheet.

- 3. Write your JAMB registration numbers on the question paper in the space provided at the top of page 1.
- 4. Attempt all questions.
- 5. The use of calculator and or similar electronic devices is NOT allowed

ENGLISH

- 1. An autobiographical novel is ______
- A. A novel written about another novelist
- B. A true account of a novelist's life by himself
- C. A novel in which the novelist draws mainly on materials from his own life
- D. A novel using the 'I' pronoun
- 2. Plot in prose fiction is best defined as:
- A. The cause and effect sequence of events
- B. The brief summary of events
- C. The central event
- D. The subject matter of a novel
- 3. Which of these is not true about unity of action in a novel?
- A. Action may be unified through a single main character
- B. Action may be unified by being set in one place
- C. Action maybe unified by many characters
- D. Action may be unified by having one consistent point of view
- 4. Action in a novel is best defined as:
- A. The summary of the novel's story
- B. What the characters do or say in the novel

- C. The numerous sub plots of the novel put together
- D. The totality of all the episodes in a novel leading to the conclusion
- 5. Suspense in a novel means _____
- A. the postponement of the hero's death till the last possible moment
- B. the intense emotions that the author conveys
- C. the inconclusive end of a novel
- D. when we are curious about what happens next in a novel
- 6. A realistic novel is one in which the characters are ____.
- A. real
- B. historic
- C. just of above average intelligence
- D. the types that we meet in everyday life
- 7. Theme is best defined as:
- A. The subject matter of a novel or play
- B. central idea in a play or novel
- C. The point of view in that novel
- D. The sum total of all the characters experiences
- 8. What figure of speech does the following quotation contain? Life's but a walking shadow (Macbeth) A. A metaphor
- B. An image
- C. A synecdoche
- D. An allusion
- 9. Which of the following statement is most true about poetry?
- A. the meaning of words is more important than their sounds
- B. the sounds of words are more important than their meanings
- C. the sound of words is often more important than their meanings
- D. sounds and meanings of words are of Little consequence
- 10. Which of these bests define exposition in drama?
- A. the author's own general introduction to the play

- B. the author's early exposure of his dramatis personae to conflict
- C. introduction to the characters and the general problem with which the play deals D. the first performance of the play on stage
- 11. When the speaker in a poem cannot be identified with the poet, that speaker is called:
- A. a persona
- B. a dramat is persona
- C. a soliloquist
- D. a ventriloquist
- 12. Which of these definitions best describes a lyric?
- A. A short poem
- B. a short poem in which the poet himself is speaking
- C. a poem expressing a personal idea, feeling or mood
- D. a poem divided into stanzas

Read the following extract from a poem to answer questions 13 and 14.

Among rocks, I am the loose one, among arrows, I am the heart among daughters, I am the recluse, among sons, the one who dies young

- 13. What kind of repetition is used in the extract?
- A. anaphora
- B. single word repetition
- C. line repetition
- D. phrase repetition
- 14. Apart from emphasis, what other effect does the repetition have?
- A. makes the poem rhythmical
- B. makes the poem tedious
- C. makes the poem monotonous
- D. makes the poem exhilarating
- 15. The following line from poem Western wind, when will thou bl ow? Is an example of:
- A. rhetorical question
- B. caesura
- C. alliteration
- D. nature imagery
- 16. Identify the odd one out of these types of imagery.
- A. visual imagery

- B. tactile imagery
- C. synaesthesia
- D. literal imagery
- 17. When a poet uses mainly soft vowel sounds in a poem, the texture of the poem is:
- A. mellifluous
- B. harsh
- C. assonantal
- D. neutral
- 18. The sounds in the following lines may be described as an example of: The moan of doves in immemorial elms, and murmuring of innumerable bees
- A. Alliteration
- B. Euphony
- C. Cacophony
- D. A mixture of all above
- 19. The man dies in him who keeps silent in the face of tyranny". This statement can be described as:
- A. Metaphoric
- B. Literal
- C. Tragic
- D. a simile
- 20. The literary technique in which a reader is taken to the past of a current action is known as:
- A. rewinding
- B. fast-forward
- C. flashback
- D. repetition

PHYSICS

- 21. A piece of rubber 1 Dem long stretches 6mm when a load of 100N is hung from it What is the area stretched if the young modulus is $50N/m^2$.
- A. 60m²
- B. 150 m²
- C. 33.33 m²
- D. 15 m²
- 22. To determine the weight of an object you would _____.
- A. use a balance
- B. use a spring balance
- C. find the force necessary to give it a certain acceleration
- D. use none of these methods
- E. use any of these methods

23. A gas at pressure PN/m and temperature 27°c is heated to 77°c at constant volume. The new pressure is A. 0.85PN/m B. 0.86PN/m C.1.16PN/m D. 1.18PN/m E. 2.85PN/m
24. Two lamp rated 40w and 220w, each are connected in series. The total power dissipated in both lamps is A. 10w B. 20w C. 40w D. 80w E. none
25. A magnet is moved through a coil of wire. The emf produced in the wire depends on
A. the number of turns in the coil B. the strength of the magnet C. the speed at which the magnet is moved D. all of the above E. none of the above
26. A potential difference of 6v is used to produce a current of 5A for 200s through a heating coil. The heat produced is A. 4800cal B. 6000cal C. 2400j D. 240kcal E. 600j
27. Two boys are communicating with each other by stretching a string passing through a hole punched in the bottom of each of the two tin cans. The physical principle employed is that sound travels

A. mainly through air

liquids

air

B. fainter in stretched string

E. none of the above is correct

C. faster through gasses than in solids and

28. The hatch door of a submarine has an

area of 0.5 m². The specific gravity of sea

water is 1.03. Assume that $g = 10 \text{m/s}^2$ and

neglect the atmospheric pressure. The force

D. with greater ease through a string than in

- exerted by the sea water on the hatch door at a depth of 200m is
- A. $1.03 \times 10^5 \,\mathrm{N}$
- B. $1.03 \times 10^4 \text{ Nm}^2$
- C. $2.6 \times 10^3 \text{ Nm}^2$
- D. 2.06 x10¹¹ N
- E. $1.03 \times 10^3 \text{ N}$.
- 29. When equal weights of iron and water are subjected loan equal supply of heat, it is found that the piece of iron becomes much hotter than water after a short time because
- A. The specific heat or iron is higher than water
- B. Iron is in solid form
- C. water is in liquid form
- D. the specific heat of water is higher than that of iron
- E. the specific heat of iron is infinite
- 30. The speed of light in is 3.0×10^8 m/s. its speed in glass having a refractive index of
- 1.65 is _____.
- A. $1.82 \times 10^8 \text{m/s}$
- B. $3 \times 10^8 \text{m/s}$
- C. $6.0 \times 10^8 \text{m/s}$
- D. $1.82 \times 10^{2} \text{m/s}$

BIOLOGY

- 31. Which of the following is an excretory organ in flatworms?
- A. Malpighian tubules
- B. Kidney
- C. Nephridium
- D. flame cells
- 32. Which of the following is not an excretory organ in mammals?
- A. Anus
- B. Liver
- C. Lung
- D. Skin
- 33. Which of the following is not freshwater?
- A. Brackish water
- B. Puddle
- C. Pond
- D. River
- 34. Which of the following is a special organ for oxygen absorption in aquatic animals?
- A. Air bladder
- B. Lung

- C. Spiracle
- D. Gills
- 35. Which of the following is not a feature of Marshes?
- A. High oxygen content
- B. Water-logged soil
- C. Low light intensity
- D. Abundant saprophytic bacteria
- 36. Drought resistant plants are called?
- A. Hydrophytes
- B. Xerophytes
- C. Mesophytes
- D. Bryophytes
- 37. Primary succession on a terrestrial habitat is associated with _____.
- A. Bare soil
- B. Abandoned farmland
- C. Primary forest
- D. Secondary forest
- 38. Which of the following may not result in overcrowding?
- A. Limited space
- B. Scarcity of food
- C. Increased birth rate
- D. Tight immigration conditions
- 39. Which is the function of mitochondria? A. Gives cell rigid shape B. Site for energy release
- C. Manufactures protein
- D. Contains cell sap
- 40. Which is not a feature of animal cells?
- A. Presence of centrosome
- B. Absence of cell wall
- C. Presence of few, small vacuoles
- D. Presence of plastids

CHEMISTRY

- 41. Which of the following in a molar solution would be the pro duct conductor of electricity?
- A. ammonia
- B. sodium chloride
- C. hydrochloric acid
- D. zinc tetraoxosulphate (VI)
- 42. How many moles of oxygen atoms are presented in 4.0g of the gas. (O=16) A. 0.25

- B. 0.50
- C. 1.00
- D. 4.00
- 43. In the periodic table, the entire element within the same group has the same
- A. atomic number
- B. number of electrons in the nuclei of their atoms
- C. number of electron available for bonding
- D. number of isotopes
- 44. How many atoms are there in 0.3mole of an element? {Avogadro's constant

 $=6.0 \times 10^{23}$)

- A. 2.0×10^{24}
- B. 1.8 x10²⁴
- C. 1.8 x10²³
- D. 3.0×10^{22}
- 45. Milk exposed to air for a few days tastes sour due to the presence of _____.
- A. ethanoic acid
- B. Citric acid
- C. lactic acid
- D. tartaric acid
- 46. A substance which produces the hydroxonium ion as the only positive ion when dissolved in water is
- A. an acid
- B. an acid salt
- C. a normal salt
- D. a base
- 47. Petrol can be obtained from fuel oil by A. hydrolysis
- B. Hydrogenation
- C. Dehydration
- D. cracking
- 48. Which of the following is not true of metals?
- A. they are good conductors of heat and electricity
- B. they are ductile and malleable
- C. their oxides are basic
- D. they can be used as insulators
- 49. Which of the following is not a property of chlorine?
- A. it is greenish yellow and has a choking smell
- B. it bleaches litmus paper

C. it is a liquid at room temperature and pressure

D. it has a higher density than air

50. If calcium has atomic number 20 and mass number 40, the constituent of the atom is

A. 20 protons, 10 neutrons and 10 electrons

B. 20 protons, 20 neutrons and 20 electrons

C. 20 protons, 20 neutrons and no electrons

D. 10 protons, 10 neutrons and 20 electrons

ANSWERS TO ABU 2017/2018 POSTUTME EXAM

ENGLISH

1. B 2. A 3. D 4. B 5. D 6. D 7. B 8. A 9. A

10. C 11. A 12. C 13. A 14. A 15. C 16. C

17. C 18. B 19 A 20. C

PHYSICS

21. C 22. B 23. C 24. E 25. A 26. B 27. D

28. E 30. A

BIOLOGY

31. D 32.A 33.D 34. B 35.A 36. 37. 38. D

39. B 40

CHEMISTRY

41. D 42.A 43. C 44.C 45.A 46.A 47. D 48. D

49. C 50.B

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2015/2016 POST-UTME SCREENING EXERCISE

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ENGLISH

COMPREHENSION PASSAGE

Tony spends most of in spare time at his medium-sized farm located on the fringes of the capital City along Okigwe road, the farm haven for the lover of retreat. It occupies an undulating land, which stretches for a good distance. The farm enjoys a modest canopy formed by palm trees, banana branches and coconut trees. The barbed wired fence bends harmoniously with the surrounding lush of the green foliage. There on the heart of the farm.

Tony raises small animals that include rabbits, goats and pigs. He ensures that animals particularly the pigs that form the main Naira earners are always in peak condition This Tony does by seeing to it that they receive then regular clinical attention Similarly, they always insist that the pens are cleaned and sanitized. Not surprisingly, the droppings from the animals are constantly carted away by enthusiastic farmers They believe that the application should significantly boost their own harvest This in part explains why there has been a good hassle for the droppings from Tony's farm in some other ways, the patronage is emblematic Some of the fanners simply I earn through their inner thought to make a statement.

Whenever Tony is out of his station, the manager follows religiously the strictly gamin already established

- 1. The second sentence suggests that Tony's farm is good _____.
- A. for religious camping and meditation
- B. for escaping from city noise
- C. as a rendezvous for layers
- D. as a location for serious meeting
- 2. From the description of the farm and surrounding,
- 3. "They received their regular clinical attention" This means that
- A. die animals are taken to the clinic regularly
- B. the animals are under constraint observation
- C. the animal is are given regular vaccination
- D. the animals have a doctor at the clinic
- 4. The farmers who use animal droppings from Tony's farm think that _____.
- A. their harvest with increase
- B, their crops will remain healthy
- C. their farm will be inexpensive
- D. their farm will be a model
- 5. Tony's farm manager can be described as
- A. a good imitator
- B. a religious person
- C. a sympathetic individual
- D. a responsible worker

In question 6 and 7, an idiom is underlined in each of the sentences, find under each sentence the group of words that gave the nearest meaning to the idiom.

- 6. He killed the goose that laid the golden egg when he decided to spend his capital on buying a car.
- A. sacrificed future profit to sati sly present needs
- B. was bad natured
- C. became a poor person
- D. became an important person
- 7. I laughed up my sleeve as he told the tales of his exploits.
- A. was very pleased
- B. was openly making him of him
- C. was secretly amused

D. laughed myself helplessly

In questions 8 and 9, choose the word or set of words that best lit the meaning of the sentence as a whole.

8._____the long run. we will make enough profit.

- A. In
- B. On
- C. At
- D. For
- 9. The congregation ____all sorts and conditions of men
- A. composes
- B. consists
- C. comprises
- D. conjectures

In questions 10 and 11, select the option that is nearest in meaning to the underlined word.

- 10. The effect of the new policy on the masses is rather <u>negligible</u>.
- A. unimportant
- B. useful
- C. insignificant
- D. negligent
- 11. When I returned to the cyber cafe a week later, my ticket had become <u>invalid</u>.
- A. inappropriate
- B. crippled
- C. discounted
- D. out of date.

In question 12 and 14, choose from the options given in A-D the one which is opposite in meaning to the word.

- 12. An <u>expert</u> will not be found waiting in this type of job hunting.
- A. A foreigner
- B. a jobber
- C. An applicant
- D. A novice
- 13. If you are not for <u>lesbianism</u>, it means you are for.
- A. celibacy
- B. homosexuality
- C. nun hood
- D. heterosexuality

- 14. The chairman's verdict <u>redressed</u> the injustice meted out by the secretary.
- A. corrected
- B. aggravated
- C. restored
- D. addressed

Interpretation:

- 15. My friend gambled about all day yesterday This means that he _____.
- A. was gambling with his money yesterday
- B. did nothing useful yesterday
- C. played cards for money
- D. was jumping in the field.

From the words or group of words lettered A-D, choose the option that is most nearly opposite to the underlined words

- 16. Our grandmother was very much loved for her <u>altruism</u>.
- A. benevolence
- B. selflessness
- C. kindness
- D. selfishness
- 17. The principal <u>lost</u> his usual good humour when his school was defeated in this year's cowbell mathematics quiz competition.
- A. provoked
- B. kept
- C. maintained
- D. obtained
- 18. Nkoli was the one who politely <u>declined</u> the invitation.
- A. turned off
- B. turned away
- C. accepted
- D. received
- 19. We quickly realized that a confrontation was <u>inevitable</u>.
- A. disastrous
- B. unavoidable
- C. desirable
- D. conditionally
- 20. Obiora's name was inadvertently omitted.
- A. unfortunately
- B. improperly
- C. intentionally

D. conditionally

CHEMISTRY

- 21. In the reaction between sodium hydroxide and sulphuric acid solutions. what volume of 0.5molar sodium hydroxide would exactly neutralize 10 cm³ of 1.25molar sulphuric acid?
- A. 5cm³
- B. 10cm³
- C. 20cm³
- D. 25cm³
- E. 50cm³
- 22. A small quantity of solid ammonium chloride (NH4CI) was heated gently in a test tube. The solid gradually disappear to produce a mixture of two gases later a white cloudy deposit was observed on the cooler part of the test tube. The ammonium chloride is said to have undergone
- A. distillation
- B. sublimation
- C. precipitation
- D. evaporation
- E. decomposition
- 23. Elements P, Q, R, S have 6, 11,15, & 17 electrons respectively. Therefore,
- A. P will form an electrovalent bond with R
- B. Q will form a covalent bond with S C. R will form an electrovalent bond with S
- D. O will form an electrovalent bond with S
- E. Q will form a covalent bond with R
- 24. An element X forms the following compounds with chlorine NCl₄, XC₃. XC₂ this illustrates the
- A. law of multiple proportions
- B. law of chemical proportions
- C. law of simple proportion
- D. law of conservation of mass
- E. law of definite proportion
- 25. The oxidation state of chlorine in potassium chlorate is .
- A. 11
- B. 12
- C. 13
- D. 15
- E. 17
- 26. When carbon dioxide is bubbled into limewater, a white precipitate is formed. If

the passage of the gas is continued, the precipitate disappears. The reasons for this is

- A. calcium carbonate is formed which on reaction dissolves
- B. calcium hydrogen carbonate is precipitated and then dissolves
- C. calcium carbonate is formed which on reaction with further carbon dioxide forms soluble calcium hydrogen carbonate
- D. concentration of solution has occurred with the deposition of calcium hydroxide
- E. the solution has become saturated and solid carbon dioxide has been deposited
- 27. The following reactions are stages in important industrial processes
- (i) $N_{2(q)} + 0_2 2 NH_{3(q)}$ tiH is negative
- (ii) $2S0_{2(g)} + 0_{2(g)} 2S0_{3(g)} \Delta H$ is negative
- (iii) $N_{2(g)} + O_{3(g)} 2NO_{(g)} \Delta H$ is positive. Which of the above forward reactions is favoured by
- (i) a decrease in the concentration of the pressure and
- (ii) an increase in temperature?
- A. i
- B. ii
- C. iii
- D. i & ii
- E. i & iii
- 28. Methanoic acid mixes with water in all proportions and has about the same boiling point as

water. Which of the following methods would you adopt to obtain pure water from a mixture of sand, water and methanoic acid A. fractional distillation

- B. filtration followed by aestivation
- C. neutralities with sodium hydroxide followed by distillation.
- D. neutralization with sodium hydroxide followed by filtration
- E. etherification with ethanol followed by distillation
- 29. Which of the following statements applies during the electrolysis of sodium hydroxide solution using platinum electrodes?
- A. anions are discharge at the cathode
- B. hydrogen ions are discharge at the anode
- C. the concentration of sodium hydroxide decreases at both electrode compartments
- D. the concentration of sodium hydroxide increases at the cathode only

- E. the concentration of sodium hydroxide increases at the anode only
- 30. Which of the following statement is true? When the potassium atom forms its ion
- A. it gains one electron and becomes neutral
- B. its atomic number decreases
- C. it achieves electronic configuration of argon
- D. it loses one proton
- E. it loses one neutron.

PHYSICS

- 31. The slope of a straight-line displacement time graph indicates _____.
- A. distance travelled
- B. uniform velocity
- C. uniform acceleration
- D. instant acceleration
- E. uniform speed
- 32. A ball of mass 0.5kg moving at 10m/s collides with another ball of equal mass at rest. If the two balls move off together after the impart, calculate their common velocity
- A. 0.2m/s
- B. 0.5m/s
- C. 10m/s
- D. 3m/s
- 33. How much heat is given out when a piece of iron mass 50g and specific heat capacity 460Jkg⁻¹K⁻¹ cools from 85°c to 25°c?
- A. $1.38 \times 10^6 \,\mathrm{J}$
- B. $2.53 \times 10^2 J$
- C. $1.98 \times 10^4 \text{ J}$
- D. $1.38 \times 10^3 J$
- E. 1.27x10³ J
- 34. Which of the following is not a suitable method of reducing loss of heat from a piece of hot iron?
- A. wrapping it in cotton wool
- B. painting it black
- C. placing it in a vacuum
- D. placing it in a rubber support
- E. keeping it in a closed wooden box
- 35. A bat emits a sound wave at a speed of 1650.00m/s and receives the echo 0.15s later. Calculate the distance of the bat from the reflector
- A. 8.75m
- B. 16 .60m

- C. 87 .75m
- D. 123.75m
- E. 330.00m
- 36. Which of the following is/are characteristics of sound?
- i. pitch
- ii. Loudness
- iii. Quality
- iv. Noise
- A. i only
- B. ii only
- C. i & ii only
- D. i, ii & iii only
- E. I, ii, iii and iv
- 37. An image which can be formed on a screen is said to be _____.
- A. virtual
- B. blurred
- C. inverted
- D. erect
- E. real
- 38. A ray of light is incident at an angle of 30° on a glass prism of refractive index 1.5. Calculate the angle through which the ray is minimally deviated in the prism (the medium surrounding the prism is air)
- A. 10.5°
- B. 5.5°
- C. 21.1°
- D. 38.9°
- E. 40.5°.
- 39. At which of the following distances from the lens should a s Ii de be placed in a slide projector if the focal length of the projection lens?
- A. less than f
- B. greater than 2f
- C. greater than f but less than 2f
- D. equal to f
- E. equal to 2f
- 40. What of the camera corresponds to the iris of the eye?
- A. shutter
- B. film
- C lens
- D. diaphragm
- E. focusing ring

BIOLOGY

- 41. The nucleus is considered the control organelle of a cell because it
- A. Contains the genetic material
- B. Contains the nuclear sap
- C. is bounded by the nuclear membrane
- D. is located at the centre of the cell
- 42. Regulation of blood sugar level takes place in the _____.
- A. pancreas
- B. Ileum
- C. Liver
- D. Kidney
- 43. The heart of the adult frog consists of
- A. two auricles and two ventricles
- B. one auricle and one ventricle
- C. two ventricle and two auricles
- D. one ventricle and two auricles
- 44. A group of organisms of different species living in a particular area is described as a
- A. colony
- B. community
- C population
- D. niche
- 45. A freshwater plant such as water lily can so live the problem of buoyancy by the possession of ____.
- A. parenchymatous tissue
- B. dissected leaves
- C thin cell walls of the epidermis
- D. water repelling epidermis
- 46. Soil micro-organisms are beneficial because of their involvement in
- A. photosynthesis B. translocation
- C. cycling of nutrients
- D. respiration using soil air
- 47. One of the ways in which body eel Is differ from gamete cells is in the .
- A. type of centromeres they contain
- B. number of chromosomes pair they contain
- C. type of chromatids they contain
- D. number of chromosomes they contain
- 48. In the blood transfusion, agglutination occurs when _____.

- A. white blood eel Is from two individuals meet
- B. two different antibodies meet
- C. two different antigens meet
- D. contrasting antigens and antibodies meet
- 49. The richest sources of vitamin A are
- A. palm oil and groundnut oil
- B. palm oil and carrots
- C rice and groundnut oil
- D. oranges and carrots
- 50. Yellowing of leaves is a symptom associated with deficiency of _____.
- A. iron, calcium and magnesium
- B. nitrogen, sulphur and potassium
- C. sulphur, phosphorous and iron
- D. magnesium, nitrogen and iron

ANSWERS TO ABU 2015/2016 POST UTME

- 1. B 2. A 3. C 4. A 5. D 6. A 7. C 8. A 9.C
- 10. C 11. D 12. D 13. B 14. B 15. B 16. D
- 17. B 18. C 19. D 20. C

CHEMISTRY

- 21. E 22. B 23. D 24. A 25. C 26. C 27. C
- 28. B 29. B 30. C

PHYSICS

- 31. C 32. C 33. D 34. A 35. D 36.D 37. E
- 38. B 39. C 40. D

BIOLOGY

- 41. A 42. A 43. D 44. B 45. A 46. C 47. B
- 48. D 49. B 50. B

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2014/2015 POST-UTME SCREENING

Time allowed: 1 hour

ENGLISH

Choose the correct option to till the blank spaces.

1. The guest	breakfast by	the time	the
hus arrives			

- A. shall have finished
- b. have finished
- C. had finished
- D. are finishing
- 2. You can go on I _____ what you are saying.
- A. will understand
- b. an understanding
- C. understanding
- D. waste understanding
- 3. When he was knocked on the head, he fell to the ground.
- A. fainted
- B. Unconscious
- C. collapsed
- D. noisily
- 4. It so ____hard that all the cars have stopped moving
- A. will rain
- B. rained
- C. rain
- D. rains
- 5. The principal will be going away on leave, in his absence the vice-principal will _____ school.
- A. overlook
- B. take over
- C. look after
- D. care for

Choose the word that is nearly opposite in meaning to the underlined word.

- 6. The able-bodies should take care of the
- A. feeble
- B. weak-minded
- C. Suffering
- D. soft-hearted

- 7. The chairman ordered him either to withdrew or to _____ his allegations
- a. affirm
- B. drew
- C. express
- D. complete
- 8. He shows plenty of goodwill to his neighbours, but they bear nothing except towards him.
- A. bad luck
- B. malice
- C. anger
- D. unhappiness
- 9. I supported what you said but I ____ the way you said it
- A. argued about
- B. objected to
- C. interfered with
- D. investigated
- 10. Though many of us were poor quite a few were .
- A. arrogant
- B. Prodigal
- C. Affluent
- D. Luxurious

MATHEMATICS

- 1. A fair coin is tossed ten times. What is the probability of getting at least two heads?
- A. $^{513}/_{1024}$
- B. $^{615}/_{1024}$
- C. $968/_{1024}$
- D. $^{1011}/_{1024}$
- 2. A man bought 220 mangoes at \\$5x. he sold each for 3x kobo and made a gain of \\$8. Find the value of x
- A. 2
- B. 5
- C. 10
- D. 6
- 3. Calculate the sum of infinity of

$$1 + \frac{1}{2} \pm \frac{1}{2} + \frac{1}{27}$$

- A. 0 .33
- B. 0 .67

- C. 1.5
- D. 2.5
- 4. If "P + "C2, then n can be ____.
- A. 1
- b. 2
- C. $\frac{3}{2}$
- D. 3
- 5. Express tan 22 ½ ° in the surd form
- A. $\sqrt{2}$ -2
- B. $1-\sqrt{2}$
- C. $\sqrt{2}$ -1
- D. $+\sqrt{2}+1$
- 6. If sin(x-a) = Cos(x+a), then tan is
- A. 0.8
- B. 0.75
- C. 1.0
- D. 6.28
- 7. Simplify 11011₃
- A. 10100_3
- B. 1100₂
- C. 1110₃
- D. 1011₂
- 8. A binary operation x is defined on R, the set read numbers by $xy = \sqrt{XY} = \sqrt{XY}$ for all X,Y ER. If X*(2*8)=6, find x
- A. 2
- B. 4
- C. 9
- D. 10
- 9. Find the mean deviation of 5.8, 7 and 2.
- A. 0
- B. 2
- C. 5
- D. 10
- 10. The volume of two similar solid cubes are 729cm³ and 512cm². Find the ratio of their lengths
- A. 4:3
- B. 9:7
- C. 3:2
- D. 9:8

PHYSICS

1. Which of the following is not true about semi-conductor?

- A. moving holes are equivalent to moving positive charges
- B. there are two kind of charges carrier; free electron and hole
- C. the escape of a vale nee electron from an atom produces electron holes pair of charge carrier
- D. Increase in temperature increases its electrical resistance
- 2. The minimum energy necessary to re move an electron from a given atom at infinity is called .
- A. excitation
- B. ground state energy
- C. ionizing energy
- D. binding energy
- 3. Find the de Broglie wavelength of a 0.01 kg pallet having a velocity of 10m/s and energy: 663×10^{34} js
- A. 6 .63 x 10³¹ m
- B. 6.63 x10⁻³² m
- C. 6 .63 x 10^{-33} m
- 4. A set-up transformer is designed to operate from a 25v supply. If the transformer is 80% efficient, determine the current in the primary coil when the output terminals are not connected to 240v 100w lamp
- A. 5.0A
- B. 4 0A
- C. 2.5A
- D. 2.0A
- 5. An object of mass 0.2kg and density 600kgm³ is suspended with a string so that it is immersed in paraffin of density 900kgm⁻³. Find the tension in the string
- A. 0.2 N
- B. 2.0 N
- C. 1.0 N
- D. 0.1 N
- 6. A rocket burns fuel at the rate of 20kgs⁻¹ and eject it with a velocity of 5.0 x 10³ m/s⁻¹. Calculate the thrust exerted by gas on the rocket.
- A. 1.0 x 10⁵ms⁻¹
- B. 2.0 x10⁵ms⁻¹
- C. 3.0 x105ms-1
- D. 5.0 x10⁵ms⁻¹
- 7. Which of the following pairs consist fundamental quantities only?

A. velocity and gravitational potential	A. Olecranon
B. acceleration and field strength	B. Ulna
C. momentum and work done	C. Tibia
D. moment and mass.	D. Humerus
8. One of the limitations of Thomson's model	5. The condition in which the anthers mature
of the atom is that it does not explain.	before the stigma is called
A. Small angle	A. protandry
B. stability of the atom	B. Epigyny
C. ionization process	C. Hypogyny
D. the variation of the effective atomic radius	D. protogyny
9. A wire carrying a current of IGA and 2.5m	6. In most true ferns, sporangia are grouped
length is placed in a field of flux density	into .
0.14T. what is the force of the wire if it is	A. indusium
placed at 600 to field?	B. fronds
A. 30.3N	C. son
B. 20.5N	D. Prothalis
C. 15.3N	
D. 10.5M	7. The ratio of carriers to suckers in the F2
	generation derived from a pare nta I cross at
10. In the transformer, the magnetism of the	two carriers of haemoglobins S gene is
core is repeatedly reverse by the magnetic	A. 3:1
field resulting in energy as heat. The loss is	B. 1:3
called	C. 2:1
A. Eddy current	D. 1:2
B. hysteresis loss	
C. flux linkage	8. In which part of a leguminous plant can
D. joule heating loss	bacteria like Azotobacterial be found?
	A. Spongy mesophyll
BIOLOGY	B. Root nodes
	C. stern internodes
 In bird. the following feathers possess after shaft 	D. Stem nodes
A. Quill and filo plumes	9. In a dicotyledonous stem, companion cells
B. Down and filo plumes	are found close to the
C. Covert and down	a. Endodermal cells
D. Quill and covert	B. Silver tubes
	C. Xylem vessels
2. The nutritive layer of the eye in mammals	D. Pericyclic fibres
is	10. The position occupied by an organism in a
A. refracting media	The position occupied by an organism in a food chain is referred to as
B. conjunctiva C. Cornea	A. Trophic level
D. Sclera	B. Niche Level
D. Sciera	C. Energy level
3. Ultra-filtration in the Kidney takes place in	D. feed level
the	
A. Bowman's Capsule	CHEMISTRY
B. Pelvis	
C. Loop of Henle	Natural water include the following except
D. Proximal Convoluted Tubule	A Deir washen
A Which of the fallenties because	A. Rain water
4. Which of the following bones is not a	B. Spring water
component of the fore limb?	C nure water

- D. Lake water
- 2. Which of the following methods cannot be used to remove permanent hardness from water?
- A. Adding of washing soda
- B. addition of caustic soda
- C. perutit method
- D. adding alum to water
- 3. If the solubility or sodium tetraoxosulphate (IV) at 30°c is 18g per 100g How much is this gram per kilogram?
- A. 18g per 100g
- B. 180g per 100g
- C. 180g per 1000g
- D. 180g per mg
- 4. The following are example of colloid except
- A. Milk
- B. starch in water
- C. aerosol
- D. ammonium chloride solution
- 5. The pH of the solution M, N, O, and P are
- 4,6,8 and 10 respectively. Therefore
- A. none of the solution is acidic
- B. the pH of the O made neutral by adding water
- C. the most acidic solution is P
- D. Mis the most acidic solution
- 6. Sodium chloride may be obtained from brine by .
- A. detection
- B. distillation
- C. evaporation
- D. sublimation
- 7. Oil spillage in pond and creek can be cleared by _____.
- A. Burning of the oil layer
- B. spraying with detergent
- C. spraying with common salt
- D. spraying with oil
- 8. Which of the following is a chemical compound?
- A. Soap
- B. milk
- C. urine
- D. gold

- 9. Crystallization is a separation method used
- A. where purity of the product is important
- B. Where beauty of the product is important
- C. where the product is a solid
- D. where the product cannot be destroyed by heat
- 10. Which hydroxide dissolves in water to form an alkaline
- A. Aluminium hydroxide
- B. Calcium hydroxide
- C. Copper hydroxide
- D. Iron hydroxide

ANSWERS TO ABU POST UTME 2014/2015

ENGLISH

1. A 2. A 3. A 4. B 5. C 6. A 7. A 8. B 9. B 10. C

MATHEMATICS

1. A 2. B 3. C 4. – 5. C 6. B 7. – 8. – 9. A 10. D

PHYSICS

1. D 2. C 3. D 4. A 5. C 6. A 7. D 8. B 9. A 10. B

BIOLOGY

1. C 2. C 3. A 4. C 5. A 6. B 7. C 8. B 9. B 10. A

CHEMISTRY

1. C 2. A 3. D 4. D 5. A 6. C 7. A 8. A 9. A 10.—

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2013/2014 Post-UTME SCREENING

Time allowed: 1 hour

Answer all questions: shade the answer sheet

as appropriate with HB pencil only

CHEMISTRY

- 1. A mixture contains 20cm³ of hydrogen, 35cm³ of oxygen, 15cm³ of carbon dioxide and 10 cm³ of nitrogen at S.T.P which of the following gives the mole fraction of hydrogen in this mixture?
- A. 0.02
- B. 0.16
- C. 0.20
- D. 0.25
- E. 2.0
- 2. 0.07g of a hydride of carbon occupies 56 at S .T.P when vaporized and contains 14 .29% by mass of hydrogen. The formula of the hydrocarbon is
- A. CH₄
- B. C_2H_2
- C. C₂H₄
- D. C₂H₆
- E. C₃H₈
- [C = 12, H = 1]
- 3. The pressure on 100cm³ of oxygen gas at 35°c is 750mm of Hg. What would be the volume of the gas if the pressure is increased to 1000mm of Hg without changing the temperature?
- A. 133.3 cm³
- B. 85 cm³
- C. 75 cm³
- D. 65 cm³
- E. 58 cm³
- 4. Which of the following bonds exist in crystalline ammonium chloride (NH₄Cl)?
- A. ionic and covalent
- B. ionic and co-ordinate
- C. ionic, covalent and co-ordinate
- D. covalent, co-ordinate and metallic
- E. ionic, covalent and metallic
- 5. Which of the following is a neutralization reaction? Addition of _____.
- A. nitric acid to hydrochloric acid
- B. nitric acid to sulphuric acid

- C. acid to distilled water
- D. nitric acid to sodium hydroxide
- E. sodium chloride to distilled water
- 6. In the preparation of carbon monoxide by heating ethanoic acid with concentrated sulphuric acid, the concentrated sulphuric acid acts as _____.
- A. oxidizing agent
- B. reducing agent
- C. dehydrating agent
- D. reaction medium
- E. catalyst
- 7. How many grams of methyl acetylene (propane) CH3-C=CH will completely discharge the colour of 8g of bromine? (Br=80, C = 12, H = 1)
- A. 0.5
- B. 1.0
- C. 2.0
- D. 3.0
- E. 4.0
- 8. Brass is an alloy containing copper and
- A. zinc
- B. tin
- C. aluminium
- D. silver
- E. lead
- 9. 60 cm^3 of hydrogen are sparked with 20cm3 of oxygen at 100°c and 1 atmosphere. The total volume of the residual gases is
- A. 60 cm³
- B. 10 cm³
- C. 40 cm³
- D. 30 cm³
- E. 70 cm³
- 10. If the rate of diffusion of oxygen gas is taken as 1, what will be the rate of diffusion of methane whose relative molar mass is 16?
- A. 2.0
- B. 1.8
- C. 1.4
- D. 1.0
- E. 0.5

USE OF ENGLISH

In each of questions 11 to 20, choose the
word(s) or phrase(s) which best fill(s)
the gap(s)

the gap(s)
11. The sea wave continue to the cliff or the west coast constantly. A. impair B. rub C. knock D. erode
12. The college bus was traveling at a highwhen the accident occurred. A. velocity B. acceleration C. rapidity D. speed
13. Note that only senior members of staff have the of using the toilet upstairs. A. permission B. occasion C. privilege D. habit
14. The chief priest willthe main into the cut today. A. indoctrinate B. usher C. convert D. initiate
15. Obi is noted for his attitude to his seniors at school. A. receptive B. Respectful C. respective D. respectable
16. The girl that my brother introduced to us last week is pretty ill-mannered. A. and B. but also C. as well as D. respectable
17. The police report wasto that of the eye witness.A. contraryB. inconsistent

C. different D. congruent
18. The African extended family system gives security tomembers. A. his B. her C. its D. their
19. I know Iread more but I am tired. A. may B. ought to C. would D. could
20. Insects can becometo insecticides A. immunized B. resistant C. Reticent D. immobilized
PHYSICS
21. Which of the following is a scalar quantity? A. momentum B. acceleration C. displacement D. distance E. force
22. What change in velocity would produce a body of mass 4kg if a constant force of 16N acts on it for 2s? A. 0.5m/s B. 2.0m/s C. 8.0m/s D. 32.0m/s E. 128.0m/s
23. A body accelerates uniformly from rest at the rate of 3m/s² for 8s. Calculate the distance covered by the body during the acceleration. A. 12m B. 24m C. 48m D. 72m E. 96m
24. Which of the following has the same unit as the moment of a force? A. force B. power

- C. Work
- D. momentum
- E. charge
- 25. Which of the following will reduce the frequency of oscillation of a simple pendulum?
- A. increasing the mass of the bob
- B. decreasing the mass of the bob
- C. increasing the length of the string
- D. decreasing the length of the string
- E. increasing the amplitude of oscillation
- 26. A barometer can be used in determining the length of a
- i. mountain
- ii. Depth of a mine
- iii. Dew point.

Which of the following is/are correct?

- A. i, ii, iii
- B. ii and iii only
- C. i and iii only
- D. i and ii only
- E. iii only
- 27. Which of the following colours of surfaces will radiate heat energy best?
- A. red
- B. white
- C. black
- D. yellow
- E. blue
- 28. A gas which obeys Charles law exactly has a volume of 283cm³ at 10°c. What is its volume at 30°c?
- A. 142 cm³
- B. 293 cm³
- C. 303 cm³
- D. 566 cm³
- E. 849 cm³
- 29. A real image of an object formed by a converging lens of focal length 15cm is 3 times the size of the object. What is the distance of the object from the lens?
- A. 30cm
- B. 25cm
- C. 20cm
- D. 15cm
- E. 10cm
- 30. How far from a cliff should a boy stand to hear the echo of his clap 0.9s later? (speed of sound in air= 330m/s)

- A. 36.67m
- B. 74.25m
- C. 148.50m
- D. 297.00m
- E. 366.67m

MATHEMATICS

- 31. Find n if $34_n = 100112$
- A. 5
- B. 6
- C. 7
- D. 8
- 32. The radius of a circle is given as 5cm subject to an error of 0.1 cm. What is the percentage error in the area of the circle?
- A. $\frac{1}{25}$
- B. $^{1}/_{4}$
- C. 4
- D. 25
- 33. What is the value of x satisfying the equation $4^{2x}/4^{3x} = 2$
- A. -2
- B. $-\frac{1}{2}$
- C. 1/2
- D. 2

34. If
$$x = 3 - \sqrt{3}$$
, find $x^2 + \frac{36}{x^2}$

- A. 9
- B. 18
- C. 24
- D. 27
- 35. Solve the equation y^2 11y + 24 = 0
- A. 8, 3
- B. 64, 9
- C. 6, 4
- D. 9, -8
- 36. A man invested a sum of ₩280.00 partly at 5% and partly at 4%. If the total interest is ₩12.80 per annum, find the amount invested at 5%.
- A. ₩14.00
- B. ₩120.00
- C. ₩140.00
- D. N160.00
- 37. Ice forms on a refrigerator ice box at the rate of 4 .06 g per minute after 1 minute. If initially there were 2g of ice, find the mass of ice formed in 5 minutes

A. 19.5	44. Which of the following is not likely to be
B. 17.0	found in the cell of a ripe tomato fruit?
C. 14.5	A. plastids
D. 12.5	B. chlorophyll
20. Ohtain a manimum valva af tha five ation	C. cellulose cell wall
38. Obtain a maximum value of the function	D. mitochondrion
$f(x) = x^3 - 12x + 11$	E. mineral salts
A5	AF Compain and he defined as diffusion of
B2	45. Osmosis can be defined as diffusion of
C. 2 D. 27	A stome and molecules through a membrane
D. 27	A. atoms and molecules through a membrane
39. Two perfect dice were thrown together.	to an area of higher concentration B. water molecules for a dilute solution to a
Determine the probability of obtaining a total	concentrated solution across a permeable
score of 8.	membrane
A. ¹ / ₁₂	C. water molecules from area of high
B. ⁵ / ₃₆	concentration to an area of low
C. ¹ / ₆	concentration
D. ⁷ / ₃₆	D. water molecules from a dilute solution to a
D. 736	concentrated solution through a semi-
40. The probability of an event P is % while	permeable membrane
that of another event Q is $^{1}/_{6}$. If the	E. perspiration and excretion
probability of both P and Q is 1 In, what is the	Li perspiration and exerction
probability of either P or Q?	46. The movement of diaphragm is a
A. $^{1}/_{96}$	characteristic of gaseous exchange in
B. ¹ / ₈	A. insect
C. ⁵ / ₆	B. Fish
D. 11/ ₁₂	C. toad
	D. mammal
BIOLOGY	E. plants
41. Which of the following organelles is used	47. In cellular respiration, energy is stored in
for locomotion in paramecium?	the form of
A. pseudopodium	A. adenosine diphosphate (ADP)
B. irichocyst	B. adenosine mono phosphate (AMP)
C. cilium	C. adenosine triphosphate (ATP)
D. pellicle	D. heat energy
E. contractile vacuole	E. electrical energy
42. Which of the following is not true of the	48. The medium in which dissolved nutrient
nucleus of a living cell? It contains	are transported in the body of vertebrates is
A. Chromosomes	called
B. nucleus	A. latex
C. nucleoplasm	B. urine
D. chromatids	C. cell sap
E. ribosomes	D. blood
	E. haemoglobin
43. The cell membranes consists of	•
A. carbohydrates and lipids	49. Which of the following structures of the
B. vitamins and proteins	leaf contains air?
C. lipids and proteins	A. guard cell
D. water and sugar	A. guara cen
or mater and bagar	B. palisade layer
E. starch and cellulose	
=	B. palisade layer

50. Which of the following organs is specially adapted for gaseous exchange in aquatic organisms?

A. lungs

B. trachea

C. gills

D. tracheoles

E. Alveoli

ANSWERS TO ABU POST UTME 2013/2014 EXAM

1. D 2. 3. C 4. C 5. D 6. C 7. C 8. A 9. E 10. D

USE OF ENGLISH

11. D 12. D 13. A 14. D 15. D 16. B 17. A

18. C 19. B 20. B

PHYSICS

21. C 22. C 23. E 24. C 25. C 26. E 27. C

28. C 29. C 30. C

MATHEMATICS

31. A 32. A 33. B 34. C 35. A 36. D 37. D

38. D 39. B 40. C

BIOLOGY

41. C 42. D 43. C 44. B 45. D 46. D 47.C

48. D 49.C 50.C

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2012/2013 POST-UTME SCREENING

Time allowed: 1 hour
1. Dimension of absolute viscosity is A. ML ⁻¹ T ⁻¹ B. MLT ⁻¹ C. ML ⁻¹ T D. MLT
2. Turbulent flow generally occurs for cases involvingA. highly viscous fluidB. very narrow passagesC. very slow-motionD. none of these
3. Forces acting on a particle setting in fluid areforces. A. gravitational and buoyant B. centrifugal and buoyant C. gravitational or centrifugal buoyant drag D. external, drag and viscous
4. Which of the following equations applies at terminal viscosity? A. mg - V - U = 1 B. mg - V - U = 0 C. V + mg - U = ma D. U - ma + mg = 0
5. Which of the following is not a high viscous fluid? A. kerosene B. glue C. grease D. glycerine
6. Mercury is an ideal barometric fluid mainly due to itsA. high densityB. low compressibilityC. low capillary actionD. very low vapour pressure
7. Dimension of surface tension is A. FL ⁻¹ B. F ⁻¹ L C. FL ⁻²

D. F⁻² L

- 8. Which of the following has the smallest least effect on the solubility of a so lute into the solvent?
- A. nature of the solute
- B. nature of the solvent
- C. temperature
- D. pressure
- 9. Which of the following is insensitive to changes in pressure?
- A. Heat of vaporization
- B. melting point
- C. heat of fusion
- D. both B and C
- 10. Vapour pressure of water at 100°C is about____bar.
- A. 0.1013
- B. 1.013
- C. 10.13
- D. 101.3
- 11. If w is the mode and z is the median of the following set of numbers:
- 2.4. 2.1. 1.6. 2 .6, 2.6, 3.7, 2.1 and 2.8. then (3w, 2z) is
- A. (2 6, 2 5)
- B. (21, 25)
- C.(78,50)
- D. (52, 50)
- 12. A trapezium has two parallel sides of length 6cm and 8cm. if the area is 42cm, find the distance between the parallel sides.
- A. 6cm
- B. 7cm
- C. 8cm
- D. 5cm
- 13. An arc of a cycle of length 22cm subtends an angle of $3y^{\circ}$ at the centre of the circle. Find the value of y, if the radius of the circle is 7cm.
- A. 30°
- B. 60°
- C. 120°
- D. 150°
- 14. Find the locus of a point which moves such that its distance from the line y = 3 is a constant k.

A.
$$y = 3 + k$$

Download
B. y = 3 - k C. y = 3 + k D. y = k- 3
15. From the following list of type vi mutation. Identify the one that is hereditary. A. genetic mutation B. somatic mutation C. germinal mutation D. gametic mutation
16. Which of these would not be a limiting factor in photosynthesis? A. O ₂ B. CO ₂ C. chlorophyll D. light
17. In a cell digestive enzyme mostly occur in
A. ribosome B. lysosome C. mitochondria D. plastids
18. Which of the following factors is not associated with aquatic habitat? A. temperature B. light intensity C. humidity D. turbidity
19. Terrestrial organisms which are capable of maintaining their body temperatures constant within fairly close limits are referred to as A. thermoclines B. protothemes C. poikilotherms D. eurytherms
20. Which of the following statements is not true of a climax vegetation? A. is ecological phenomenon B. is as table community C. eliminates competition D. results from succession
21. The causative organisms of sleeping sickness is the A. trypanosome B. plasmodium C. vibrio bacterium

D. penicillin

From the words lettered A to D, choose the word or group of words that best completes each of the following sentences. 22. Adamu is a very good friend on _____ I can rely on. A. who B. whose C. which D. whom 23. I ought the letter by now A. to be sent B. to have being sent C. to have sent D. to have to send 24. Edna's fever was so acute that she an injection A. had to have B. had to had C. must have D. ought to have 25. I know of a cow_has only three legs. A. whom B. which C. who D. of which 26. Ranti: This is not my key. Ayo: Then Is it? A. of which B. whose C. who's D. whom 27. There was no meat in the market_Ada bought some fish. A. so B. unless C. since D. whereas 28. The police vehicles raced full speed with their sirens blaring. A. on B. with C. at

29. Tutu liked to read detective novels to

take his mind_his worries.

D. in

A. off

3. away C. out of	C. its vapour pressure equals the atmospheric pressure
D. from	D. its volume is slightly increased
30. I don't know what to do with these	37. If the relative rate of diffusion of a gas is
children. They are always fighting.	0.25 and that of CI under the same condition
A. themselves	is 0.20. calculate the relative molecular mass
B. myself	of the gas.
C. one another	A. 22.7
D. each of them	B. 45.4
	C. 68.1
31. Wale couldn't have said a thing like that	D. 90.8
A. could he?	B. 30.0
B. did he?	38. The following molecules contain hydrogen
C. can he?	bonding EXCEPT
D. would he?	A. ammonia
D. Would Her	
22. If also had known also wouldn't have	B. ethanoic acid
32. If she had known, she wouldn't have	C. hydrogen fluoride
come	D. water
A. would she?	2 5 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7
B. wasn't it?	39. If 20 cm ³ of distilled water is added to
C. wouldn't it?	80cm ³ of 0.50mol/dm ³ HCI solution, the new
D. couldn't she?	concentration of the acid is
	A. 0.10 mol/dm ³
33. The separation of o ii and water with	B. 0.20mol/dm ³
different boiling points can best be achieved	C. 0.40mol/dm ³
oy:	D. 2 .00mol/dm3
A. fractional distillation	
B. decantation	40. What is H_2O_2 acting as in the equation?
C. evaporation	$H_2O_2 + 2Fe^{2+} \rightarrow 2Fe^{3+}$
D. using a separating funnel	A. oxidizing agent
	B. reducing agent
34. Calculate the minimum volume of oxygen	C. an acid
that is required for the complete combustion	D. a base
of a mixture of 20cm ³ of CO and 25cm ³ of	
hydrogen.	41. A 12V battery supplying a current of 20A
A. 45 cm ³	was used to melt 1.5kg of ice at $0\degree$ C.
B. 22.5cm ³	Calculate the time required if the latent heat
C. 20 cm ³	of fusion of ice is 336 x 10J/kg
D. 10 cm ³	A. 35.0 min
	B. 3.5 min
35. An increase in temperature causes an	C. 76 min
ncrease in the pressure of a gas because	D. 21.0 min
there is an increase in the	
A. average velocity of the gas molecules	42. The light from the sun reaches the earth
B. number of collisions between the gas	mainly by
molecules	A. convection
C. density of the gas molecules	B. conduction
D. free mean path between each molecule	C. radiation
and the other	D. reflection
36. A liquid begins to boil when	43. One valid assumption of the kinetic
A. its vapour pressure is equal to the vapour	theory of gases is that
pressure of its solid at a given temperature	A. the molecules are in random motion and
B. molecules start escaping from the surface	the number of collisions is constant

B. the number of molecules increases with the pressure C. the molecules of the gas are all identical and are very small in size D. the number of molecules increases with temperature 44. An astronomical telescope is said to be in normal adjustment when the _____. A. eye is accommodated B. focal length of the objective lens is longer than that of the eye piece C. final image is at the near point of the eye D. final image is at infinity 45. Which of the following parts of a cell is living? A. cell wall B. calcium oxalate C. food vacuole D. mitochondria 46. Cells without an organized nucleus are called . A. heterokaryote B. eukaryote C. prokaryote D. synkaryote 47. The sites for energy transfer within a cell are known as A. Golgi apparatus B. parenchyma C. mitochondria D. nucleolus 48. Food and dissolved oxygen pass from the water directly into the amoeba by a process called A. transport B. diffusion C. fission D. transpiration 49. Which one of these functions is not performed by the nervous system? A. receive sensory input from internal and external environment B. digestion C. integration D. response to stimuli

50. In man, gas exchange occurs in the

A. heart

- B. white blood cells
- C. lungs
- D. kidney

ANSWERS TO ABU 2012/2013 POST-UTME

- 1. A 2. C 3. C 4. A 5. B 6.A 7. A 8. D 9. A
- 10. D11. C 12. A 13. B 14. C 15. B 16. A
- 17. A 18. D 19. C 20. C 21. A 22. D 23. C
- 24. C 25. B 26. B 27. A 28. A 29. A 30. A
- 31. A 32. A 33. D 34. D 35. B 36. A 37. B
- 38. C 39. C 40. A 41. D 42. D 43. D 44. B
- 45. D 46. B 47. C 48. D 49. B 50. C

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2010/2011 POST-	-UIME SCREENING
Time allowed: 1 hour	B. thus C. that
USE OF ENGLISH	D. then
From the words or group of words lettered A to D, choose the word or group of words that best complete each of the following sentences 1of the five boys was ab le to show me the way to the zoo.	8. He has not been seen by his parentlast month. A. since B. for C. getting D. since over
A. none B. neither C. any D. some	9. The, came here last week.A. handsome tall young manB. young tall handsome manC. tall handsome young manD. young handsome tall man
2. Thisbe David's handwriting; I know his handwriting well enough.A. mayB. willC. oughtD. can't	10. The activities marking the golden jubiled celebration of the club were with a party A. rounded off B. rounded over C. rounded through D. rounded up
3 thing she had in the room were thrown out. A. so few B. the few C. all few D. very few	Choose the option that is most nearly opposite in meaning to the underlined word and that will, at the same time, correctly fill the gap in the sentence.
4. There are five boys A. of which two were beaten B. whom two were beaten C. two of whom were beaten D. of whom two of them were beaten	11. To encourage productivity we must reward industry and laziness. A. with hold B. withdraw C. punish D. oppose
5. This is the mantold me the story. A. whom I said B. who I said C. said that he D. who I said he	12. It is quite <u>customary</u> to introduce the guest speaker but to insult him. A. illegally B. impolite C. unusual D. useless
6. Many students find mathematicsthan English.A. difficultB. too difficultC. very difficultD. more difficult	13. I <u>encourage</u> my younger brothers to take on law as a profession while my sister from doing so. A. warned B. dissuaded C. persuaded
7. It is such a bad place I will never dream of going there another time.A. therefore	D. helped

14. Olu was able to kindle the fire which my father had to later. A. switch B. extinguish C. destroy D. ignite
15. It is curious how Bayo can be so <u>carefree</u> in his ways when his brother is so A. meticulous B. eccentric C. easy-going D. indifferent

Choose the option that best explain the underlined idiomatic expression in each sentence.

- 16. Mr. John has always managed to <u>keep his head oppression</u>.
- A. keep his head above water while swimming
- B. known the technique of swimming
- C. stay out of financial difficulty
- D. he is trying to pay up it borrowed money
- 17. The men eventually get their own back on their oppression.
- A. strike
- B. have their revenge on
- C. beat up
- D. abuse
- 18. He <u>went off the rails</u> as soon as he heard of his failure in the last examination.
- A. became annoyed
- B. wept bitterly
- C. became disorganized
- D. lost consciousness
- 19. She is <u>eating her heart</u> out for a sailor who is away at the sea.
- A. long for
- B. quarrelling with
- C. fuming about
- D. hating
- 20. You can't make bricks without straw.
- A. use a straw for making bricks
- B. afford not to have all the necessary materials
- C. seek a leader
- D. erect a brick without straws

PHYSICS

- 21. Which of these statements is correct about cathode rays. They are fast moving
- A. atoms
- B. neutrons
- C. electrons
- D. ions
- 22. Which of the following has the highest surface tension?
- A. soapy water
- B. cold water
- C. warm water
- D. salt water
- 23. A truck traveling with a velocity of 40m/s applies the brakes and comes to a halt after 20s. what is the distance travelled by the truck before coming to a halt
- A. 40m
- B. 800m
- C. 400m
- D. 10m
- 24. If the linear expansivity of a metal rod is 4×10^{-5} per °C, what will be the new length of the rod if it is heated from 15°C to 95°C from its original length of 20cm.
- A. 0.064cm
- B. 0.64cm
- C. 20.64cm
- D. 20 064cm
- 25. 44kJ heat was used in raising the temperature of 2kg of paraffin oil from 360k to 370k. calculate the specific head capacity of paraffin oil.
- A. 2.2J/kg/K
- B. $2.2 \times 10^{3} \text{J/kg/K}$
- C. $2.2 \times 10^5 \text{ J/kg/K}$
- D. 220J/kg/K
- 26. A simple machine overcomes a load 4000N when a force of 200N is applied. If the velocity ratio of the machine is 25, calculate the efficiency of the machine.
- A. 1.25%
- B. 80%
- C. 125%
- D. 0.8%
- 27. A long sighted person is to read a book held at a distance of 20cm from the eyes.

Which of the following will the person require to read the book with ease? A. nothing B. concave lens C. convex lens D. concave mirror	34. When a red blood cell is placed in water, the process of water movement is A. osmosis B. diffusion C. imbibition D. active transport
28. An object at the bottom of a pool of liquid 10 m deep is seen by an observer as if it is at 8m deep. What is the refractive index of the liquid? A. 0.25 B. 0.20 C. 1.25 D. 0.8 29. Which of the following has the highest surface tension? A. soapy water B. cold water C. warm water	35. The enzyme contained in bile is A. trypsin B. lipase C. ptyalin D. lactase 36. Which of these factors in the blood is responsible for blood clotting? A. fibrinogen B. heparin C. plasma D. red blood cells 37. Fat soluble vitamins are stored in
D. salt water 30. The emf developed in a circuit is directly proportional to the rate of change of magnetic flux. The above was a finding from	A. liver B. spleen C. pancreas D. skin
A. Maxwell B. Faraday C. Ampere D. Lenz	38. The unit of the nervous system is A. neuron B. axon
BIOLOGY	C. dendrite D. myelin sheet
31. Which of these is a sense organ? A. spleen B. mouth C. eye D. heart 32. What is the function of the vas deferens in the male reproductive organ?	39. Moulting involves the A. shedding of exoskeleton in insects in order to replace with better ones B. shedding of exoskeleton in insects in order to grow new ones C. shedding of exoskeleton in insects in response to seasonal changes D. none of the above
A. transports sperm only B. transports urine and sperm C. transports urine only D. none of the above 33. Food materials manufactured in plants are transported through the	40. Pepsin is a digestive enzyme which breaks A. sucrose into glucose and fructose B. carbohydrates into simple sugars C. protein into peptones D. fats into glycerol and fatty acids
A. xylem B. phloem C. cambium	CHEMISTRY
D. cortex	41. If the rate law obtained for a given reaction is given as rate = $K[X]^n[Y]^m$. what is the overall order of the reaction?

A. nm

B. n/m

C. n + m

D. n - m

42. A molecular formula shows _____ in a molecule.

A. the elements present

B. the number of atoms of each element

C. cations and anions

D. chemical symbols and number of atoms

43. Give the total mass of copper in 1 gram of copper (II) sulphate [Cu = 40, S = 32, 0=16]

A.025g

B.050q

C.10g

D.25g

44. $^{114}_{55}Cs \rightarrow ^{A}_{Z}E + ^{4}_{2}\alpha$. Find the value of A and Zin the equation above.

A. 119, 53

B. 110, 57

C. 11 0, 53

D. 110, 58

45. How many moles of H_2 molecules are needed to convert 5mole of O_2 molecules to water?

A. 5 mole H₂

B. 10 mole H₂

C. 15mole H₂

D. 20 mole H₂

46. $^{226}_{88}Ra \rightarrow {}^{4}_{Z}Rn + \alpha.$ What is the value of x in the nuclear reaction above?

A. 220

B. 222

C. 226

D. 227

47. When naphthalene on heating changes from solid state directly to the gaseous state, it undergoes

A. sublimation

B. evaporation

C. combustion

D. decomposition

48. Which of the following is an electrolyte?

A. alcohol

B. sodium ethanoate

C. solid potassium hydroxide

D. mercury

49. The equation ${}^{14}_{7}N + {}^{4}_{2}He \rightarrow {}^{17}_{8}O + {}^{1}_{1}P$

A. nuclear fusion

B. nuclear fission

C. artificial radioactivity

D. nuclear fission using positron

50. Which of the following is a general method of preparing acids?

A. direct combination of constituent elements

B. double decomposition involving a salt solution

C. reaction between an anhydride and water

D. reaction between a base and an amphoteric oxide

E. decomposition of hydroxides followed by neutralization

ANSWERS TO ABU 2010/2011 POSTUTME

ENGLISH

1. A 2. A 3. B 4. A 5. A 6. D 7. C 8. A 9. C

10. A 11. C 12. C 13. B 14. B 15. A 16. C

17. C 18. C 19. A 20. B

PHYSICS

21. C 22. B 23. A 24. D 25. B 26. B 27. C

28. C 29. B 30. B

BIOLOGY

31. C 32. A 33. B 34. A 35. --- 36. A 37. A

38. A 39. B 40. C

CHEMISTRY

41. C 42. D 43. A 44. C 45. B 46. C 47. A

48. B 49. C 50. A

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